Course Outline

MATS3004
Polymer Science and Engineering 1
Materials Science and Engineering
Science
T2, 2020
1. Staff

<table>
<thead>
<tr>
<th>Position</th>
<th>Name</th>
<th>Email</th>
<th>Consultation times and locations</th>
<th>Contact Details</th>
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</thead>
<tbody>
<tr>
<td>Course Convenor and Lecturer</td>
<td>Dr. Damia Mawad</td>
<td><a href="mailto:damia.mawad@unsw.edu.au">damia.mawad@unsw.edu.au</a></td>
<td>Room 246, School of Materials Science and Engineering (Building E10)</td>
<td>Phone: 9385 6642</td>
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2. Course information

Units of credit: 6  
Pre-requisite(s): None  

2.1 Course summary

The course is divided into 4 sections:

**Polymer Chemistry and Synthesis:** raw materials and synthesis of polymers; monomers, homopolymers, copolymers; basic organic chemistry and applied polymer chemistry; free radical polymerization, kinetics and mechanism; ionic and condensation polymerization, kinetics and mechanism.

**Polymer Physics and Characterisation:** chain dimensions; polymer thermodynamics; polymer solution behaviour; characterisation techniques in solution and solid state using spectroscopy, chromatography and colligative properties.

**Polymer morphology and physico-chemical properties:** amorphous and crystalline states; multicomponent systems; iso-free volume theory; mechanical, physical and chemical properties in relation to bulk morphology; online lab component that complements the theory learned in class with focus on the properties of polymers.

**Commodity and specialty plastics:** compositions and fabrication; additives in plastics; commercial manufacturing processes and applications

2.2 Course aims

To develop an understanding of the fundamentals in polymer science and engineering covering:

1. the synthesis and characterisation of polymers
2. the relation between chemical structure, polymer morphology and physico-chemical properties
3. technological developments in fabrication and processing techniques

The course objectives are:

- To introduce the students to polymers, their synthesis, reaction mechanism and kinetics (Lectures 1-5 + tutorial)
- To provide the students with an understanding of polymer behaviour in the solid and solution state; as well as characterisation techniques commonly used in polymer science (Lectures 6-10 + tutorial)
• To provide the students with basic knowledge of the morphology of polymers in the solid state, amorphous and crystalline. Particular emphasis will be on the interplay between morphology and physico-chemical properties (Lectures 11-14 + tutorial)
• To introduce polymer technology including processing, applications and manufacturing of commodity polymers (Lectures 15-16)

2.3 Course learning outcomes (CLO)
At the successful completion of this course you (the student) should be able to:
1. Use fundamental polymer chemistry to explain and predict the synthesis of polymers as well as the resultant structure and properties
2. Identify suitable characterisation techniques based on polymer solubility and chemical structure
3. Predict/interpret the behaviour and properties of polymers as a function of their morphology and composition
4. Evaluate technological developments in commodity and advanced polymers

2.4 Relationship between course and program learning outcomes and assessments

<table>
<thead>
<tr>
<th>Course Learning Outcome (CLO)</th>
<th>LO Statement</th>
<th>Program Learning Outcome (PLO)</th>
<th>Related Tasks &amp; Assessment</th>
</tr>
</thead>
<tbody>
<tr>
<td>CLO 1</td>
<td>Use…</td>
<td>1.1, 1.3</td>
<td>Assignment &amp; Mid-term exam</td>
</tr>
<tr>
<td>CLO 2</td>
<td>Identify…</td>
<td>1.5, 2.3, 3.3, 3.6</td>
<td>Assignment, Mid-term exam, Lab reports</td>
</tr>
<tr>
<td>CLO 3</td>
<td>Predict/interpret …</td>
<td>2.3, 3.3, 2.2</td>
<td>Lab reports &amp; Final exam</td>
</tr>
<tr>
<td>CLO 4</td>
<td>Evaluate…</td>
<td>2.1, 2.3 2.2</td>
<td>Final exam</td>
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3. Strategies and approaches to learning

3.1 Learning and teaching activities
(Based on UNSW Learning Guidelines)
• Students are actively engaged in the learning process.

It is expected that, in addition to attending classes, students will read, write, discuss, and engage in analysing the course content.

• Effective learning is supported by a climate of inquiry where students feel appropriately challenged.

Students are expected to be challenged by the course content and to challenge their own preconceptions, knowledge, and understanding by questioning information, concepts, and approaches during class and study.

• Learning is more effective when students’ prior experience and knowledge are recognised and built on.

Coursework, tutorials, assignments, laboratories, examinations, and other forms of learning and assessment are intended to provide students with the opportunity to cross-reference these activities in a meaningful way with their own experience and knowledge.
• Students become more engaged in the learning process if they can see the relevance of their studies to professional and disciplinary contexts

The course content is designed to incorporate both theoretical and practical concepts, where the latter is intended to be applicable to real-world situations and contexts.

Lectures: The core concepts will be taught in lectures, students will have access to the lectures notes before class for annotation during the lecture. Students will be engaged in the learning process through class discussions and problem-solving questions independently and working together with partners and groups.

Labs: Experimental techniques and procedures will be taught online through recorded laboratories practicals. Students will be able to reflect on the experiments and learn to process data through the lab reports after class.

3.2 Expectations of students

• Students must attend at least 80% of all classes with the expectation that students only miss classes due to illness or unforeseen circumstances
• Students must read through lecture notes and lab sheets prior to class
• During class, students are expected to engage actively in class discussions
• Students should work through lecture, tutorial and textbook questions
• Students should read through the relevant chapters of the prescribed textbook.
• Students should complete all assessment tasks and submit them on time.
• Students are expected to participate in online discussions through the Moodle page
4. Course schedule and structure

This course consists of 52 hours of class contact hours. You are expected to take an additional 94 hours of non-class contact hours to complete assessments, readings and exam preparation spread throughout the term.

<table>
<thead>
<tr>
<th>Week</th>
<th>Topics</th>
<th>Activity</th>
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</table>
| 1    | Introduction to Polymers  
Morphology/Types of Polymers  
Radical Polymerization |          |
| 2    | Radical / Ionic Polymerization  
Condensation  
Tutorial | Formative quiz |
| 3    | Polymer Thermodynamics I  
Polymer Thermodynamics II  
Characterization of Polymers I: Polymers in Solutions |          |
| 4    | Characterization of Polymers II: Molecular Weight, Light Scattering  
Characterization of Polymers III: Spectroscopy  
Tutorial | Assignment |
| 5    | Elastomers  
Revision | Mid-term exam |
| 6    | Amorphous Polymers  
Tutorial: Base Knowledge Required for the Lab Practicals | Laboratory |
| 7    | Crystalline Polymers | Laboratory |
| 8    | Mechanical Properties: Introductory level  
Tutorial | Laboratory |
| 9    | Polymer Processing I  
Polymer Processing II | Laboratory |
| 10   | Tutorial  
Revision |          |
5. Assessment

5.1 Assessment tasks

<table>
<thead>
<tr>
<th>Assessment task</th>
<th>Description</th>
<th>Weight</th>
<th>Due date</th>
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<tbody>
<tr>
<td>Formative quiz:</td>
<td>In-class quiz to assess how the students are understanding the concepts taught within weeks 1-3. This will provide students feedback on how they are progressing in the course before the Census date</td>
<td>0%</td>
<td>Week 2</td>
</tr>
<tr>
<td>Assignment:</td>
<td>Assesses the content taught in lectures 1-8</td>
<td>20%</td>
<td>Week 4</td>
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<tr>
<td>Mid-term exam:</td>
<td>In-class exam covering the content taught in lectures 1-10</td>
<td>35%</td>
<td>Week 5</td>
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<td>Group laboratory reports:</td>
<td>Laboratory activities will involve use of FTIR, Raman, XRD, DSC for analysis of polymers Results and discussion of these labs to be compiled into two reports.</td>
<td>10%</td>
<td>1 final lab report</td>
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<tr>
<td>Final exam:</td>
<td>Exam will be 2 hours in duration Topics taught in lectures 11-16</td>
<td>35%</td>
<td>Final exam period</td>
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Further information

UNSW grading system: [https://student.unsw.edu.au/grades](https://student.unsw.edu.au/grades)

1.2 Assessment criteria and standards

Assignment criteria will be available on the course Moodle page.

Students who fail to achieve a score of at least 40% for the overall exam component (i.e., mid-session exam and final exam marks combined), but achieve a final mark >50% for the course, will be awarded a UF (Unsatisfactory Fail) for the course.

Please refer to the UNSW guide to grades: [https://student.unsw.edu.au/grades](https://student.unsw.edu.au/grades)

1.3 Submission of assessment tasks

2. UNSW operates under a Fit to Sit/ Submit rule for all assessments. If a student wishes to submit an application for special consideration for an exam or assessment, the application must be submitted prior to the start of the exam or before an assessment is submitted. If a student sits the exam/ submits an assignment, they are declaring themselves well enough to do so. Information on this process can be found here: [https://student.unsw.edu.au/special-consideration](https://student.unsw.edu.au/special-consideration). Medical certificates or other appropriate documents must be included. Students should also advise the lecturer of the situation.
• Unless otherwise specified in the task criteria, all assignments must be uploaded via Moodle prior to the due date for submission.
• Assignments/lab reports submitted after the due date for submission will receive a 10% of maximum grade penalty for every day late, or part thereof.
• Students who have a disability that requires some adjustment in their teaching or learning environment are encouraged to discuss their study needs with the course coordinator prior to, or at the commencement of, their course, or with the Equity Officer (Disability) in the Equity and Diversity Unit: https://student.unsw.edu.au/disability. Early notification is essential to enable any necessary adjustments to be made.
• Rules governing conduct during exams are given at: https://student.unsw.edu.au/exam-rules

5.4. Feedback on assessment

Assignments: Feedback will be given two weeks after submission of the assignment and take the form of the mark for the assignment, overall comments on how the class performed, any common areas that were not answered correctly. Additionally, personal feedback and how each student performed may be given.

Lab reports: Students will receive their mark and individualised feedback on the areas they excelled at and which areas of the reports that were not answered correctly. Feedback will be provided through Moodle, two weeks after submission.

Midsession exams: Students will receive their marked exams indicating what questions were answered correctly and incorrectly. Overall comments and worked solutions may be provided to the class.

Final exam: Students will receive their final mark.

6. Academic integrity, referencing and plagiarism

Referencing is a way of acknowledging the sources of information that you use to research your assignments. You need to provide a reference whenever you draw on someone else's words, ideas or research. Not referencing other people's work can constitute plagiarism.

Further information about referencing styles can be located at https://student.unsw.edu.au/referencing

Academic integrity is fundamental to success at university. Academic integrity can be defined as a commitment to six fundamental values in academic pursuits: honesty, trust, fairness, respect, responsibility and courage. At UNSW, this means that your work must be your own, and others’ ideas should be appropriately acknowledged. If you don’t follow these rules, plagiarism may be detected in your work.

Further information about academic integrity and plagiarism can be located at:

• The Current Students site https://student.unsw.edu.au/plagiarism, and
• The ELISE training site http://subjectguides.library.unsw.edu.au/elise/presenting

The Conduct and Integrity Unit provides further resources to assist you to understand your conduct obligations as a student: https://student.unsw.edu.au/conduct.

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7. Readings and resources

Recommended Textbook*

Other useful textbooks
- M. Chanda. Introduction to Polymer Science and Chemistry, 2nd Ed. CRC Press, 2013

In addition, notes and selected reference material will be issued in lectures.

* Assistance is available from the Library: info.library.unsw.edu.au/web/services/teaching.html

8. Administrative matters

School Office: Room 137, Building E10 School of Materials Science and Engineering
School Website: http://www.materials.unsw.edu.au/
Faculty Office: Robert Webster Building, Room 128
Faculty Website: http://www.science.unsw.edu.au/

9. Additional support for students

- The Current Students Gateway: https://student.unsw.edu.au/
- Academic Skills and Support: https://student.unsw.edu.au/academic-skills
- Student Wellbeing, Health and Safety: https://student.unsw.edu.au/wellbeing
- Disability Support Services: https://student.unsw.edu.au/disability-services
- UNSW IT Service Centre: https://www.it.unsw.edu.au/students/index.html
- Special Consideration: https://student.unsw.edu.au/special-consideration